

Synthesis of the Hydroazulene Ring
System of Guanacastepene

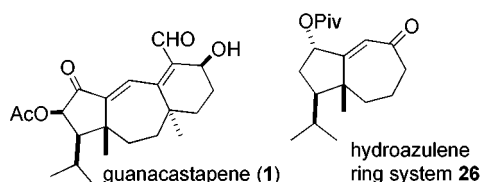
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ABSTRACT

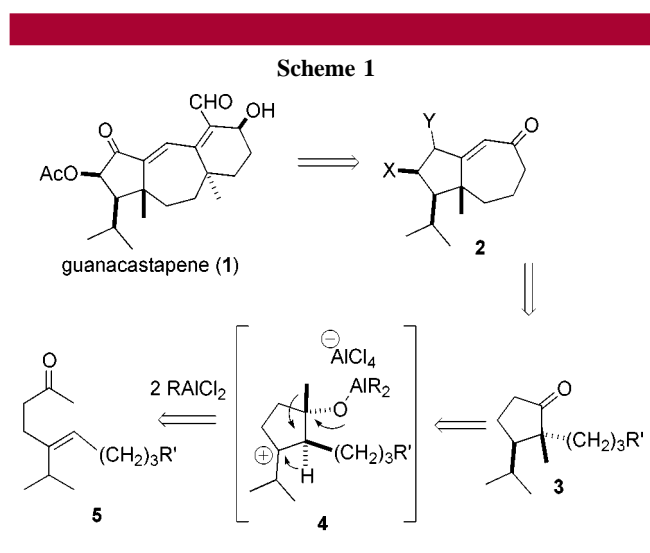


A 12-step synthesis of **26**, the functionalized hydroazulenone ring of guanacastepene (**1**), has been completed using the EtAlCl_2 -initiated cyclization of γ,δ -unsaturated ketone **13** to construct 2,2,3-trisubstituted cyclopentanone **14**, the palladium-catalyzed coupling of vinylmagnesium bromide with enol triflate **17** to prepare triene **21**, and olefin metathesis of triene **21** to form the key hydroazulene **20**.

The novel, diterpene antibiotic guanacastepene (**1**), which was isolated from an unidentified fungus growing on the tree *Daphnopsis americana*, shows excellent activity toward methicillin-resistant *Staphylococcus aureus* and vancomycin-resistant *Enterococcus faecium*.¹ Further biological studies indicated that **1** has moderate activity against Gram-positive bacteria, poor activity against Gram-negative bacteria, and hemolytic activity against human red blood cells.² The structure of **1** was determined by X-ray crystallography by Clardy and co-workers.¹ The NMR spectrum indicates that the cycloheptene ring of **1** exists as a mixture of two slowly equilibrating conformers.

Retrosynthetic analysis suggested that the cyclohexene ring of guanacastepene (**1**) could be synthesized from hydroazulenone **2** by methylation and Robinson annulation. The order of steps will have to be determined experimentally because of the conformational flexibility of the cycloheptene ring of **1** and, presumably, **2** as well. Elaboration of the functionality on the cyclohexane and cyclopentane rings will then complete the synthesis. A variety of sequences can be

envisioned to construct the cycloheptenone of hydroazulenone **2** from cyclopentanone **3**.



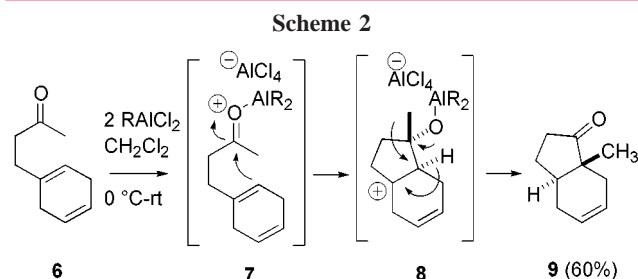
(1) Brady, S. F.; Singh, M. P.; Janso, J. E.; Clardy, J. *J. Am. Chem. Soc.* **2000**, *122*, 2116–2117.

(2) Singh, M. P.; Janso, J. E.; Luckman, S. W.; Brady, S. F.; Clardy, J.; Greenstein, M.; Maiese, W. M. *J. Antibiot.* **2000**, *53*, 256–261.

(3) (a) Snider, B. B.; Kirk, T. C. *J. Am. Chem. Soc.* **1983**, *105*, 2364–2368. (b) Snider, B. B.; Cartaya-Marin, C. P. *J. Org. Chem.* **1984**, *49*, 153–157.

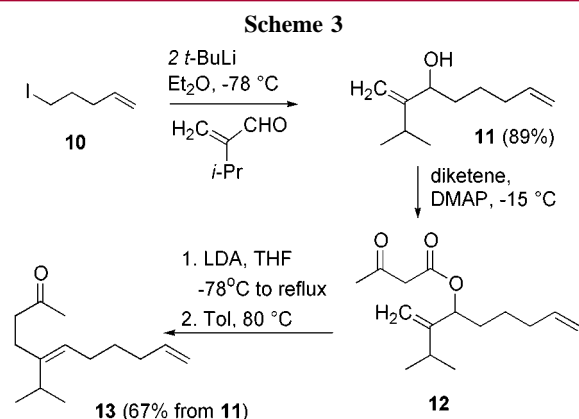
We thought that cyclopentanone **3** could be prepared stereospecifically by the EtAlCl_2 -initiated cyclization of γ,δ -unsaturated ketone **5**. Several years ago we reported a new method for cyclopentanone annulation by treatment of γ,δ -

unsaturated ketones with excess AlCl_3 .³ For instance, treatment of dienone **6** with 2 equiv of AlCl_3 generated the very electrophilic complex **7**, which cyclized to give zwitterion **8** with the bulky R_2AlO group cis to the ring fusion hydrogen. Concerted 1,2-hydride and methyl shifts afforded 60% of trans-fused hydrindenone **9**, which was elaborated to an 11-oxosteroid intermediate (see Scheme 2).



This method has been used mainly to make fused ring systems but also should be suitable for the construction of 2,2,3-trialkylcyclopentanone **3** by cyclization of γ,δ -unsaturated ketone **5** to give zwitterion **4**, which should undergo concerted 1,2-hydride and methyl shifts to give **3** stereospecifically. We chose to carry out this reaction on **13** with a side chain containing a synthetically versatile terminal double bond since the successful cyclization of **6** indicated that isolated double bonds are compatible with the acidic conditions required for this cyclization.^{4–6}

4-Pentenyllithium, prepared by halogen–metal exchange⁷ of 5-iodo-1-pentene (**10**)⁸ with 2 equiv of *t*-BuLi in THF at $-78\text{ }^\circ\text{C}$, was added to 2-isopropylacrolein⁹ to afford 89% of allylic alcohol **11**. Reaction of **11** with diketene and DMAP provided acetoacetate **12** (see Scheme 3). Enolate

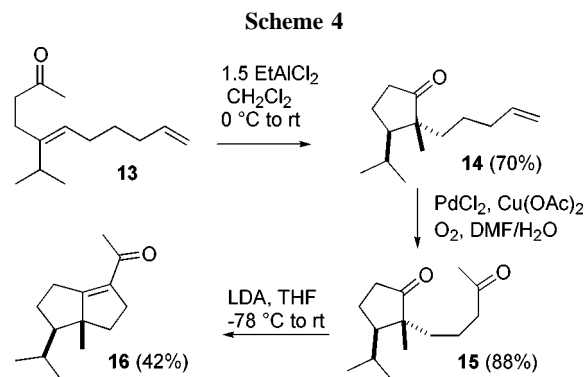


accelerated Carroll rearrangement by the procedure of Wilson¹⁰ was effected by treating **12** with 2 equiv of LDA

(4) Related cyclopentanones have been prepared by trapping the enolate formed by addition of an isopropylcuprate to 2-methyl-2-cyclopentenone with MVK⁵ or allylic halides.⁶ This approach is unlikely to be successful with less reactive alkyl halides.

in THF at $-78\text{ }^\circ\text{C}$ and heating at reflux to give the β -keto acid. Decarboxylation by heating in toluene at $80\text{ }^\circ\text{C}$ provided dienone **13** in 67% yield from alcohol **11**.

We were delighted to find that treatment of γ,δ -unsaturated ketone **13** with 1.5 equiv of EtAlCl_2 in CH_2Cl_2 at $0\text{ }^\circ\text{C}$ with gradual warming to room temperature over 24 h gave 70% of cyclopentanone **14** as the only cyclic product (see Scheme 4). The stereochemistry of **14** was established by NOE



studies on the cyclopentanone prepared by phenylselenylation, oxidation, and elimination. The allylic methine hydrogen adjacent to the isopropyl group at δ 2.45 showed an NOE to the methylene protons on the side chain at δ 1.51–1.34 but not to the methyl singlet at δ 1.04.¹¹

Initially, we planned to form the cycloheptenone by an intramolecular aldol reaction, which has been used successfully to form related hydroazulenones.¹² Diketone **15** was prepared in 88% yield by Wacker oxidation of **14** with PdCl_2 , $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ and O_2 (1 atm) in DMF.¹³ Unfortunately, no reaction occurred on treatment of diketone **15** with pyrrolidine at $80\text{ }^\circ\text{C}$ ^{12a} or KOH in MeOH .^{12b} Aldol reaction was eventually achieved by reaction of diketone **15** with LDA in THF at -78 to $0\text{ }^\circ\text{C}$ to give 42% of acetylhydropentalene **16** rather than the desired hydroazulenone.

We then unsuccessfully explored sequences involving nucleophilic addition to the carbonyl group of cyclopentanone **14**. Enolization was the only reaction even with unhindered nucleophiles such as *n*-BuLi and CeCl_3 . The carbonyl group is very hindered by the three adjacent

(5) Duhamel, P.; Dujardin, G.; Hennequin, L.; Poirier, J.-M. *J. Chem. Soc., Perkin Trans. 1* **1992**, 387–396.

(6) Piers, E.; Renaud, J.; Rettig, S. J. *Synthesis* **1998**, 590–602.

(7) Negishi, E.-i.; Swanson, D. R.; Rousset, C. J. *J. Org. Chem.* **1990**, 55, 5406–5409.

(8) Prepared in 96% yield by reaction of 5-bromo-1-pentene with NaI in acetone at reflux for 2 h. Only 59% of **11** was obtained from halogen–metal exchange with 5-bromo-1-pentene.

(9) Riehs, G.; Urban, E. *Tetrahedron* **1996**, 52, 1221–1230.

(10) (a) Wilson, S. R.; Price, M. F. *J. Org. Chem.* **1984**, 49, 722–725.

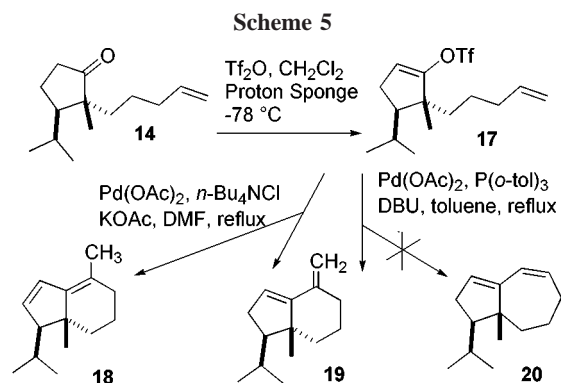
(b) Snider, B. B.; Beal, R. B. *J. Org. Chem.* **1988**, 53, 4508–4515.

(11) The NOE study was carried out in the analogous series prepared by addition of *n*-BuLi rather than 4-pentenyllithium to isopropylacrolein.

(12) (a) Ravi Kumar, V. T.; Swaminathan, S.; Rajagopalan, K. *J. Org. Chem.* **1985**, 50, 5867–5869. (b) Money, T.; Wong, M. K. C. *Tetrahedron* **1996**, 52, 6307–6324.

(13) Smith, A. B., III; Cho, Y. S.; Friestad, G. K. *Tetrahedron Lett.* **1998**, 39, 8765–8768.

substituents, whereas the α -position is unhindered. We decided to take advantage of the facile enolization of **14** by exploring Pd-insertion reactions of enol triflate **17** (see Scheme 5).



Reaction of cyclopentanone **14** with Tf_2O and Proton Sponge afforded enol triflate **17** in 86% yield. A Heck reaction on enol triflate **17** with $\text{Pd}(\text{OAc})_2$, $\text{P}(o\text{-tol})_3$, and DBU in toluene at reflux gave only the 6-exo product **19** and none of the desired 7-endo product hydroazulene **20**. Rigby has reported endo-selective intramolecular Heck reactions using Jeffery phosphine-free conditions: $\text{Pd}(\text{OAc})_2$, $n\text{-Bu}_4\text{NCl}$, and KOAc in DMF.¹⁴ Using these conditions we obtained a mixture of **19** and the isomerized diene **18**. Endo selective intramolecular Heck reactions have also been described by Genêt using TPPTS (*m*-sulfonated triphenylphosphine) in aqueous solution.¹⁵ However, treatment of triflate **17** with $\text{Pd}(\text{OAc})_2$ and TPPTS in water and CH_3CN gave only the 6-exo product **19**.

We then decided to prepare hydroazulene **20** by olefin metathesis of triene **21**. Formation of triene **21** proved to be quite challenging as a result of the facility of the intramolecular 6-exo Heck reaction that formed **19**. For instance, treatment of triflate **17** with 5 equiv of tributylvinyltin, Pd_2dba_3 (0.02 equiv), and tri-2-furylphosphine (TFP) (0.04 equiv) in THF gave a 9:1 mixture of Heck product **19** and triene **21** even though these conditions have been reported to accelerate Stille coupling.¹⁶ We then tried to prepare triene **21** without competing Heck reaction by Cu-catalyzed coupling of vinylmagnesium bromide with triflate **17**.¹⁷ Unfortunately, we obtained only traces of triene **21**, in agreement with previous reports that cuprate coupling with enol triflates can be problematic.¹⁸ Finally, we concluded

(14) Rigby, J. H.; Hughes, R. C.; Heeg, M. J. *J. Am. Chem. Soc.* **1995**, *117*, 7834–7835.

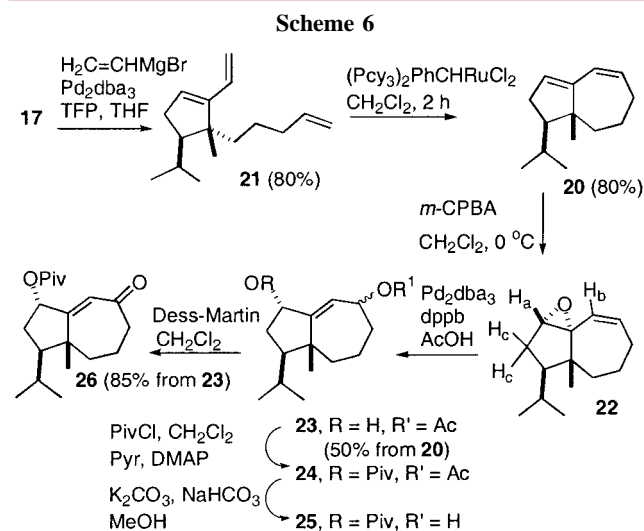
(15) (a) Lemaire-Audoire, S.; Savignac, M.; Dupuis, C.; Genêt, J.-P. *Tetrahedron Lett.* **1996**, *37*, 2003–2006. (b) Bombrun, A.; Sageot, O. *Tetrahedron Lett.* **1997**, *38*, 1057–1060.

(16) Farina, V.; Krishnan, B. *J. Am. Chem. Soc.* **1991**, *113*, 9585–9595.

(17) McMurry, J. E.; Scott, W. J. *Tetrahedron Lett.* **1980**, *21*, 4313–4316.

(18) (a) Lipshutz, B. H.; Vivian, R. W. *Tetrahedron Lett.* **1999**, *40*, 2871–2874. (b) Lipshutz, B. H.; Elworthy, T. R. *J. Org. Chem.* **1990**, *55*, 1695–1696. (c) Hirama, M.; Nakamine, T.; Itô, S. *Tetrahedron Lett.* **1988**, *29*, 1197–1198.

that Pd-catalyzed coupling of vinylmagnesium bromide with enol triflate **17** should be faster than the intramolecular Heck reaction (see Scheme 6).¹⁹



We were pleased to find that treatment of triflate **17** with vinylmagnesium bromide and $\text{Pd}(\text{Ph}_3\text{P})_4$ in THF at reflux gave a 9:1 mixture of triene **21** and Heck product **19**. The formation of the Heck product can be completely suppressed using catalyst conditions optimized for Stille coupling.¹⁶ Treatment of **17** with vinylmagnesium bromide (5 equiv), Pd_2dba_3 (0.02 equiv), and tri-2-furylphosphine (TFP) (0.04 equiv) in THF at room temperature gave triene **21** in 80% yield. Olefin metathesis²⁰ of triene **21** with bis(tricyclohexylphosphine)benzylidineruthenium dichloride (Grubbs' catalyst, 2×0.1 equiv) in CH_2Cl_2 for 2 h cleanly afforded the desired hydroazulene **20** in 80% yield. The use of CH_2Cl_2 , rather than benzene, as solvent was crucial for the success of this reaction. Starting material was still present and significant amounts of byproducts had formed after reaction for 2 d in benzene.

Treatment of hydroazulene **20** with *m*-CPBA at 0°C in CH_2Cl_2 and saturated aqueous NaHCO_3 gave epoxide **22** as a single stereoisomer. A 1D NOESY spectrum of epoxide **22** with irradiation of the epoxide hydrogen H_a at δ 3.26 shows a larger cross peak to the alkene hydrogen H_b at δ 5.50 than to the cyclopentane methylene hydrogens H_c at δ 2.06 and 1.33. The distances calculated by conformational searching with MM2 minimization in **22** are $\text{H}_a\text{--H}_b = 2.80$ Å and $\text{H}_a\text{--H}_c = 2.62$ and 2.69 Å. The calculated distances in the stereoisomer in which epoxidation occurred from the more hindered β -face are $\text{H}_a\text{--H}_b = 3.37$ Å and $\text{H}_a\text{--H}_c = 2.54$ and 2.83 Å. In this isomer the NOE between the epoxide hydrogen H_a and the alkene hydrogen H_b should be much smaller than those to the cyclopentane methylene hydrogens H_c .

(19) Murahashi, S.-I.; Yamamura, M.; Yanagisawa, K.-i.; Mita, N.; Kondo, K. *J. Org. Chem.* **1979**, *44*, 2408–2417.

(20) (a) Schneider, M. F.; Junga, H.; Blechert, S. *Tetrahedron* **1995**, *51*, 13003–13014. (b) Fürstner, A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3012–3043. (c) Yet, L. *Chem. Rev.* **2000**, *100*, 2963–3007.

Treatment of epoxide **22** under Deardorff's conditions,²¹ (Ph₃P)₄Pd and AcOH in THF at 65 °C, afforded 30% of acetoxy alcohol **23** as 1:1 mixture of acetate stereoisomers. The yield increased to 50% with Pd₂dba₃, dppb, and AcOH in THF at 65 °C.²² We were unable to open the vinyl epoxide with triphenylsilanol by the Trost procedure.²¹ Deardorff reported the exclusive formation of the cis acetoxy alcohol from cyclopentadiene monoepoxide.²⁰ It is not clear why there was a loss of stereocontrol with epoxide **22**, but this is of no concern since this stereocenter is lost on oxidation to the enone.

Acetoxy alcohol **23** was elaborated to hydroazulenone **26** in 85% overall yield by protection of the alcohol with pivaloyl chloride, DMAP, and pyridine in CH₂Cl₂ to give pivaloate **24**, selective hydrolysis of the acetate using K₂CO₃ and NaHCO₃ in MeOH to give alcohol **25**, and

(21) (a) Deardorff, D. R.; Myles, D. C.; MacFerrin, K. D. *Tetrahedron Lett.* **1985**, *26*, 5615–5618. (b) Deardorff, D. R.; Myles, D. C. *Organic Synthesis, Collective Volume 8*; Wiley & Sons: New York, 1993; pp 13–19.

(22) Trost, B. M.; Ito, N.; Greenspan, P. D. *Tetrahedron Lett.* **1993**, *34*, 1421–1424.

oxidation with the Dess–Martin reagent to give hydroazulenone **26**.

In conclusion, we have completed the synthesis of the functionalized hydroazulenone ring system of guanacastepene in 12 steps using the EtAlCl₂-initiated cyclization of γ,δ -unsaturated ketone **13** for the preparation of 2,2,3-trisubstituted cyclopentanone **14**, the palladium-catalyzed coupling of vinylmagnesium bromide with enol triflate **17** to prepare triene **21**, and olefin metathesis of triene **21** to form the key hydroazulene **20**. We are currently exploring methods to elaborate the cyclohexene ring and modify the functionality in the cyclopentane ring to complete the synthesis of guanacastepene.

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Supporting Information Available: Full experimental procedures for the sequence leading to **26**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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